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53. ChemQuest 17.

Name: _____ Date:

_____ Hour: _____

Information:

Separating Charges.

Examine Figure 1

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Answers where there are three pairs of metal spheres that have different amounts of charge on them. The spheres in diagram C are closer than the others because they have the strongest attraction.

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Thermodynamics

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This video is designed to help you complete the molarity chemquest found in Unit 11.

Molarity

118 ChemQuest 39

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Answer Hour: _____

Information: Molarity Concentration is a term that describes the amount of solute that is dissolved in a solution.

Concentrated solutions contain a lot of dissolved solute, but dilute solutions contain only a little.

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An element exists as
4 different isotopes.
4.35% have a mass of
49.9461, 83.79%
have a mass of

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Answers
51.9405, 9.50% have a mass of 52.9407 amu, and 2.36% have a mass of 53.9389 amu. a. What is the average atomic mass for this element? b. What is the identity of this element?

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Answer Key More

Hint: add your answer to question 1 and four times your answer to question 2. $4 + 4(7) = 4 + 28 = 32$

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15. Given your answer to question 14 and the reaction you wrote in question 13, calculate the molarity of the nitrate ions (NO_3^-) in the solution. Title:

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CLASSES

(Base the answer to this question on the number of moles of propane that actually get combusted—which is your answer to part a.) 12 moles. For every mole of propane that combusts 3 moles of CO_2 are produced, so the number of moles of CO_2 that can be

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Answers
produced when 4
moles of propane
combusts = $4(3) = 12$.

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